

# Enhancing the Mechanical Properties of Alkali-Activated Mortar with Graphene Quantum Dots

## Nitikron Weingyod<sup>1,\*</sup> Kantipok Hamcumpai<sup>1</sup>, Pattharaphon Chindasiriphan<sup>1</sup>, Pitcha Jongvivatsakul<sup>1,3</sup> and Suched Likitlersuang<sup>2,3</sup>

<sup>1</sup> Center of Excellence in Innovative Construction Materials, Department of Civil Engineering, Faculty of Engineering, Chulalongkorn University,

Bangkok, 10330, Thailand.

<sup>2</sup> Centre of Excellence in Geotechnical and Geoenvironmental Engineering, Department of Civil Engineering, Faculty of Engineering, Chulalongkorn University, Bangkok 10330, Thailand.

<sup>3</sup> GreenTech Nexus: Research Center for Sustainable Construction Innovation, Faculty of Engineering, Chulalongkorn University, Bangkok, 10330, Thailand.

\*Corresponding author; E-mail address: 6670136021@student.chula.ac.th

#### **Abstract**

Graphene quantum dots (GQDs) are nanoscale carbon particles with exceptional properties, making them highly suitable for a wide range of applications. In this study, GQDs were incorporated into alkali-activated mortar (AAM) to enhance their workability and mechanical performance. The GQDs were added at dosages of 0.3%, 0.6%, 0.9%, and 1.2% by weight of fly ash. The study investigates the impact of GQDs on key properties of AAM, including compressive strength, flexural strength, porosity, environmental impact, and eco-strength efficiency. The results showed that the addition of 0.3% GQDs significantly improved mechanical strength, with a 39.1% increase in compressive strength compared to the control mixture without GQDs. The flexural strength increased by 38.8% compared to the control mix, while the porosity decreased by 4%. However, when the GQD content exceeded 0.3%, a reductions in both compressive strength and flexural strength was observed. Environmental impact analysis revealed that higher GQD content led to a reduction in carbon emissions. Additionally, the eco-strength efficiency assessment demonstrated that the mix containing 0.3% GQDs achieved the highest eco-strength efficiency among all tested mixes. This finding underscores the potential of GQDenhanced AAM for practical, sustainable applications.

Keywords: Alkali-activated material, Carbon emission, Environmental impact, Fly ash, Graphene quantum dots (GQDs)

#### 1. Introduction

In recent years, the greenhouse effect became a significant driver of extreme climate events, including droughts and rising average temperatures [1]. Cement production, accounting for approximately 7% of global CO<sub>2</sub> emissions, was one of the major contributors to anthropogenic emissions, underscoring the urgent need for sustainable alternatives within the construction industry, such as the transition to low-carbon binder materials like alkaliactivated materials (AAMs) and geopolymers. [2, 3].

The synthesis of AAMs involves the use of an aluminosilicate precursor, which reacts with an alkali activator, initiating a polymerization reaction [4]. This process results in AAM pastes with properties similar to those of cement paste. The aluminosilicate precursors are typically sourced from industrial solid wastes, including slag, fly ash, and other aluminosilicate materials [5]. These industrial by-products contribute to environmental pollution and disposal challenges, particularly fly ash from coal-fired power plants. Coal combustion contributes to 27% of global energy production and 38% of global electricity generation [6].

AAMs offer a viable alternative to Portland cement, with their binding capacity resulting from the chemical activation of industrial by-products that serve as aluminosilicate sources [7]. Unlike Portland cement-based materials, which primarily produce hydration products such as calcium silicate hydrate (C-S-H), calcium hydroxide, and ettringite, the binding capacity of



AAMs comes from two main gel phases: calcium aluminosilicate hydrate (C-A-S-H) and sodium (or potassium) aluminosilicate hydrate ((N,K)-A-S-H) [8]. Over the past few decades, AAMs have gained recognition as environmentally friendly alternatives to traditional cementitious materials [9]. Moreover, previous studies [10, 11] have demonstrated that AAM production results in lower  $CO_2$  emissions compared to traditional cement production, further reinforcing their potential as an eco-friendly alternative.

Graphene quantum dots (GQDs), characterized by crystal sizes ranging from 5 to 20 nm, were novel carbon-based materials that hold significant potential to revolutionize material engineering [12]. GQDs exhibited exceptional mechanical strength, thermal and electrical conductivity, and hydrophobic properties, owing to their nanoscale dimensions and twodimensional structure [13]. GQDs have been utilized as additives in cement-based materials to improve their environmental sustainability. The exceptional properties of GQDs make them an effective catalytic material, making them well-suited for use as an additive to enhance material properties. Win et al. [14] investigated the application of GQDs in the development of cement composites. Their results demonstrated enhancements in workability, compressive strength, and flexural strength, primarily attributed to the improved microstructural bridging mechanism, which facilitates the formation of C-S-H products. According to Raj et al. [15], GQDs were applied as an additive to enhance the performance of railway concrete sleepers. Their findings revealed that incorporating 0.3% GQDs significantly improved concrete properties, including workability, mechanical strength, and durability. Moreover, this approach reduced the CO<sub>2</sub> footprint of railway track production by 668 tons per 100 km compared to tracks manufactured using traditional ordinary Portland cement (OPC). However, research on incorporating nanoparticle additives into AAMs to evaluate their effects on workability and strength remains limited. Further investigation is needed to optimize their application and assess their potential in developing more sustainable alternatives for future industrial use.

Therefore, the objective of this manuscript is to investigate the influence of GQDs on the mechanical properties, porosity, and eco-strength efficiency of AAMs, with the aim of developing a sustainable and high-performance construction material aligned with the goals of environmental sustainability and infrastructure innovation.

## 2. Materials and procedures

#### 2.1 Materials

The materials utilized in this study include high-calcium fly ash (HCF), sodium metasilicate anhydrous ( $Na_2SiO_3$ -anhydrous), natural sand, water, and GQDs. The HCF, shown in Fig. 1, was sourced from the Mae Moh Power Plant located in the Mae Moh District, Lampang Province, Thailand. The fly ash was classified as Class C according to ASTM C618 [16], with a loss on ignition (LOI) of 0.15% and a specific gravity of 2.35. The chemical composition is presented in Table 1.

Sodium metasilicate anhydrous ( $Na_2SiO_3$ -anhydrous) was used as an alkali activator. It played a key role in dissolving aluminosilicate from the precursor, promoting the formation of a three-dimensional network structure, and enhancing the binding capacity of the AAMs, as illustrated in Fig. 2. The powdered sodium metasilicate anhydrous used in this study contained 51%  $SiO_2$ , 46%  $Na_2O$ , and 3%  $H_2O$ .

Natural sand, shown in Fig. 3, was used as a fine aggregate and conformed to ASTM C33 [17]. It had a specific gravity of 2.66, a fineness modulus of 2.85, and a water absorption rate of 0.58%.

GQDs were synthesized by CrystalLyte Co., Ltd. under patent number WO2024147021A1 for GC86s (liquid GQDs), as shown in Fig. 4. The GQDs were produced via electrochemical reduction of a saturated  $\rm CO_2$ -monoethanolamine (MEA) solution. The density of the GQD-containing MEA solution used in this study was 1,050 kg/m³.



Fig. 1 High-calcium fly ash



Fig. 2 Sodium metasilicate anhydrous



Table 1 Chemical composition of fly ash

Chemical composition	HCF (%)	
SiO <sub>2</sub>	31.5	
CaO	26.4	
$Al_2O_3$	15.0	
Fe <sub>2</sub> O <sub>3</sub>	13.8	
SO <sub>3</sub>	4.99	
MgO	2.76	
Na <sub>2</sub> O	2.27	
K <sub>2</sub> O	2.12	
TiO <sub>2</sub>	0.34	
P <sub>2</sub> O <sub>5</sub>	0.268	
ВаО	0.144	
MnO	0.12	



Fig. 3 Natural sand



Fig. 4 GQDs

## 2.2 Mixture proportions

The AAM mix proportions utilized a liquid/ash (L/A) ratio of 0.29. GQDs were introduced at dosages of 0.3%, 0.6%, 0.9%, and 1.2% by binder weight. The detailed mix proportions are presented in the Table 2.

Table 2 Mix proportions of AAM

Mix	HCF (kg/m³)	Na <sub>2</sub> SiO <sub>3</sub> (kg/m³)	Water (kg/m³)	Sand (kg/m³)	GQDs (kg/m³)
Control	790	79	252	950	-
GQDs0.3	790	79	252	950	2.4
GQDs0.6	790	79	252	950	4.7
GQDs0.9	790	79	252	950	7.1
GQDs1.2	790	79	252	950	9.5

#### 2.3 AAM mixing procedure

The mixing process for the AAM mortar consists of three main steps. In the first step, a dry mixture of HCF, sodium metasilicate anhydrous, and sand were mixed for three minutes. In the second step, water and the additive material (GQDs) were added, and the mixture was blended for three minutes. In the third step, after mixing, the fresh AAM was placed into a mold, wrapped in a plastic sheet, and cured in an oven at 80 °C for 24 hours. Subsequently, the samples are demolded and allowed to cure in ambient air at room temperature until the testing day.

## 2.4 Experimental investigations

This experimental study examines the incorporation of GQDs into AAM, with a focus on mechanical properties and environmental impact. The workability, mechanical properties, and durability were assessed through evaluations of flowability, compressive strength, flexural strength, and porosity. For the environmental evaluation, the total equivalent CO<sub>2</sub> emissions and eco-strength efficiency were determined. A total of three specimens was used for each test.

## 2.4.1 Flowability

To assess the workability of AAM, the flowability of the mortar was evaluated according to the ASTM C1437 [18]

## 2.4.2 Compressive strength

The compressive strength test was conducted in accordance with ASTM C109 [19]. The testing is performed on three  $50\times50\times50$  mm<sup>3</sup> cube specimens, cured for 7, 14, 28, and 90 days.



#### 2.4.3 Flexural strength

The flexural strength of mortar was evaluated in accordance with ASTM C348 [20], using three specimens with dimensions of  $40 \times 40 \times 160 \text{ mm}^3$  at 28 days of age.

#### 2.4.4 Porosity

The porosity was determined in accordance with ASTM C642 [21]. The test is conducted on three cube specimens measuring  $50\times50\times50$  mm<sup>3</sup>, which were cured for 28 days.

#### 2.4.5 Environmental impact

The total equivalent  ${\rm CO_2}$  emissions are calculated using Eq. (1), and the obtained values were analyzed the eco-strength efficiency (ESE). The environmental impact is assessed based on the ESE, which is calculated using Eq. (2). The equivalent  ${\rm CO_2}$  emission factors are illustrated in Table 3.

Total of equivalent  $CO_2$  emission =  $\sum_{i=1}^{n} (QM \times EFM)$  (1)

$$ESE = \frac{Compessive strength}{Total of equivalent CO_2 emission}$$
 (2)

where QM is the quantity of material (kg/m $^3$ ), EFM is the equivalent CO $_2$  emission factor of material (kg CO $_2$ -e/kg).

Table 3 Equivalent CO<sub>2</sub> emission factor

Materials	Equivalent CO2 emission factor		
	(kg CO2-e/kg)		
HCF	0.0213 [22]		
Sodium metasilicate	0.705 [22]		
Sand	0.00471 [22]		
GQDs	-0.63 [15]		
Water	0 [22]		

## 3. Result and discussion

## 3.1 Workability

Fig. 5 illustrates the flowability of AAM mortar incorporating GQDs. The results indicate that flowability improved with increasing GQD content. The GQDs1.2 mixture demonstrated the highest flowability at 113%, marking a 9% increase compared to the control mixture. This enhancement is likely due to the intrinsic hydrophilic properties of GQDs, which help improve the mortar's workability [12].

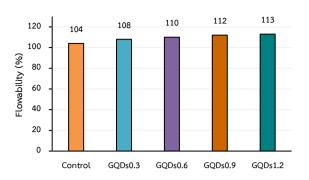


Fig. 5 Flowability of AAM

### 3.2 Compressive strength

Fig. 6 presents the compressive strength of AAMs incorporating GQDs at curing ages of 7, 14, 28, and 90 days, with values ranging from 16.4 to 33.7 MPa. The results show that adding up to 0.3% GQDs enhanced the compressive strength at all curing ages. The GQDs0.3 mixture exhibited the highest strength at 28 days, reaching 33.4 MPa an increase of 33% compared to the control mixture. This improvement is likely due to the ability of GQDs to accelerate chemical reactions, promoting the formation of C-S-H and C-A-S-H gels, which contribute significantly to strength development [12].

However, when the GQD content exceeded 0.3%, a reduction in compressive strength was observed. This decline may be attributed to the agglomeration of GQDs at higher concentrations, leading to poor dispersion and negatively affecting the composite's performance [23].

Interestingly, the compressive strength at 90 days showed only a marginal increase compared to the 28-day values across all mixtures. This is primarily because the alkali-activation process was nearly complete by 28 days of curing.

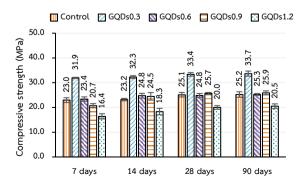


Fig. 6 Compressive strength of AAM



#### 3.3 Flexural strength

The results of the flexural strength test are shown in Fig. 7, with values ranging from 4.8 to 7.4 MPa. The data indicate that flexural strength increased with the addition of GQDs up to 0.3%. However, further increases beyond 0.3% led to a decline in strength. Notably, the GQDs0.3 mixture achieved the highest flexural strength of 7.4 MPa, representing a 40% increase compared to the control mixture. This improvement is likely due to the bridging effect of graphene derivatives, which can function similarly to microfibers, enhancing the flexural performance of the composite [14]. On the other hand, adding more than 0.3% GQDs resulted in reduced flexural strength, following a trend similar to that observed in compressive strength results. This decrease may be attributed to poor dispersion and aggregation of GQDs at higher concentrations, which can negatively affect the composite's integrity.

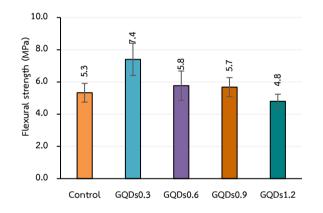


Fig. 7 Flexural strength of AAM

#### 3.4 Porosity

Fig. 8 illustrates the porosity results of AAM incorporating GQDs. Overall, the observed porosity percentage ranged from 15.2% to 19.2%. It discovered that the porosity results were significantly reduced when the GQDs was added in the system. The lowest porosity, 15.2%, was highlighted in GQDs0.3, illustrating a reduction of porosity by 4% compared to control mixture. This is probably due to the enhanced nucleation effect of the graphene derivatives, which facilitates the formation of the C-S-H phase [13]. Previous research has emphasized the capacity of graphene nanomaterials to enhance the refinement of pore distribution [24,25].

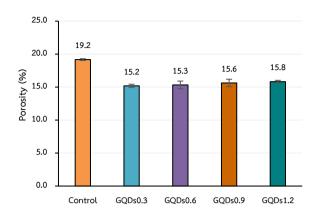


Fig. 8 Porosity of AAM

#### 3.5 Environmental impact

Fig. 9 and Fig. 10 illustrate the total embodied  $CO_2$  emissions and eco-strength efficiency of AAM incorporating GQDs, respectively. Based on the evaluation, it was observed that increasing in GQDs level addition assisted to reduce the embodied  $CO_2$  emissions [26]. During the manufacturing process,  $CO_2$  was utilized in the synthesis of GQDs, thereby classifying them as carbon-negative products. The  $CO_2$  emission factor for GQDs has been reported as -0.63 kg  $CO_2$ -equivalents per kilogram [15].

In terms of eco-strength efficiency, the GQDs0.3 mixture exhibited the highest value of 0.467 MPa/kgCO $_2$  m $^3$ , outperforming all other batches. This obviously indicates a suitable option in both strength and environmental consideration. Consequently, the GQDs0.3 mixture can be considered a suitable candidate for developing eco-friendly construction materials for practical applications.

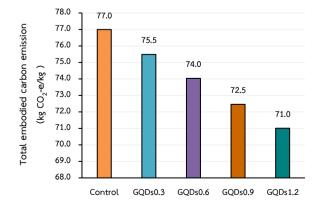


Fig. 9 Total embodied carbon emission of AAM



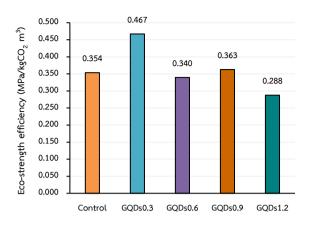


Fig. 10 Eco-strength efficiency of AAM

#### 4. Conclusions

This study explored the incorporation of GQDs in AAM at varying dosages to assess their effects on workability, mechanical properties, durability, and environmental impact. The findings indicate that adding GQDs enhanced flowability, with the highest improvement observed at increased dosages. The optimal performance was observed at a GQD dosage of 0.3%, leading to substantial improvements in mechanical properties and durability. The compressive and flexural strengths at 28 days increased by 33% and 40%, respectively, compared to the control mixture. However, exceeding 0.3% GQDs addition resulted in adverse effect on mechanical properties. This is attributed to the increased porosity caused by the incorporation of GQDs. In addition, the environmental assessment revealed that increasing GQD content reduced embodied CO<sub>2</sub> emissions, with 0.3% GQDs exhibiting the highest eco-strength efficiency. Therefore, incorporating GQDs, particularly at 0.3%, presents a promising strategy for developing eco-friendly construction materials. For future investigations on GQDs, a more comprehensive examination of their properties is warranted to evaluate their potential as suitable materials for industrial applications. Such research could facilitate the development of environmentally friendly materials and contribute to the reduction of CO<sub>2</sub> emissions.

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